Bell Ringer #7:

Socrative Room Name: LEVEL70WARRIOR

Conversion/Density

http://drmoad.weebly.com/

Agenda

Bell RingerFinish Aluminium Lab

Notes: States of Matter

Test Outline Exit Ticket

Finish Aluminium Foil Thickness Lab

Data Table

Foil Piece	Length (cm)	Width (cm)	Area (cm²)	Mass (g)	Density (g/cm³)	Volume (cm³)	Thickness (cm)
Foil #1					2.6989		
Foil #2					2.6989		
Foil #3					2.6989		

1. Calculate the area for each piece of foil. Show all calculation. Area = Length x Width. Fill in the table appropriately.

2. Calculate the volume for each piece of foil. Show all calculations. Volume = Mass/Density. Fill in the table appropriately.

3. Calculate the thickness of each piece of foil. Show all calculations. Thickness = Volume/Area. Fill in the table appropriately.

4. Calculate your percent error for this experiment and determine at least three sources of possible error.

Percent Error = $\frac{\text{(experimental value-accepted value)}}{\text{accepted value}} \times 100\%$

Kinetic Theory

- The word kinetic refers to motion
- The energy an object has because of its motion is called kinetic energy
- According to the kinetic theory, all matter consists of tiny particles that are in constant motion.
- The particles in a gas are usually molecules or atoms.

• Three states of matter are solid, liquid and gas.

Solids

- A solid is a form of matter that has a definite shape and volume.
- As a result, solids are almost incompressible; that is, it is difficult to squeeze a solid into a smaller volume.
- In addition, solids expand only slightly when heated.

Liquids

- The volume of a liquid is fixed or constant.
- Thus, a liquid is a form of matter that has an indefinite shape, flows, and yet has a fixed volume.

Gases

- Like a liquid, a gas takes the shape of its container.
- But, unlike a liquid, a gas can expand to fill any volume.
- A gas is a form of matter that takes both the shape and volume of its container.



PhET Interactive Simulations: States of Matter

• Students will observe a computer simulation of the three states of matter.

https://phet.colorado.edu/en/simulations/category/chemistry

Board Meeting:

Draw two lines and divide your board into 4 sections.

Three Sections: Draw the three states of matter.

Fourth Section: Draw the shape and volume of just one state of matter (your choice).

Test Outline: Unit 1 Test

The following topics will be covered on the test:

- Conversions
- Density
- Lab equipment
- Quantitative & Qualitative Observations
- Scientific notation
- Clues for chemical changes
- States of matter

Exit Ticket #4:

Socrative Room Name: LEVEL70WARRIOR

Scientific Notation

New Lab: Aluminium Foil Thickness

Great Value Brand Aluminium Foil Thickness = 0.00090 in

1 in = 2.54 cm

Convert to cm