Bell Ringer #40:

# Socrative Room Name: LEVEL70WARRIOR

# **Chemical Reactions Predicting Products**

http://drmoad.weebly.com/

2C2H6 + 1002 -> 1 CO2+ 10H20 C 4 C 4 H 12 O 14

# Agenda

Bell Ringer
Five Types of Reactions Notes
Demo: Elephant Toothpaste
Demo: Precipitate Formation
Predicting Products
Reaction Type Worksheet
Exit Ticket

# Five Types of Chemical Reactions

- 1. Single Replacement
- 2. Double Replacement
- 3. Synthesis
- 4. Decomposition
- 5. Combustion

## Single Replacement

$$A + BC \rightarrow B + AC$$

- One element replaces another in a compound.
- Elements and compounds are produced.

$$Zn + 2 AgNO_3 \rightarrow 2 Ag + Zn(NO_3)_2$$

$$Zn + H_2SO_4 \rightarrow H_2 + ZnSO_4$$

### Double Replacement

# $AB + CD \rightarrow AD + BC$

- Two elements switch places.
- Some types include: precipitation, gas formation, hydrolysis, neutralization

$$AgNO_3 + KBr \rightarrow AgBr + KNO_3$$

$$FeCl_3 + 3 HOH \rightarrow Fe(OH)_3 + 3 HCl$$

# **Demo: Precipitate Formation**

#### Reactants:

potassium iodide lead (II) nitrate

#### **Products:**

lead (II) iodide potassium nitrate Pb (NO3)2 DKI -> Pb I2 +2K NO3

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# 2 KI + Pb(NO<sub>3</sub>)<sub>2</sub> → 2 KNO<sub>3</sub> + PbI<sub>2</sub> V V Clear Clear in the Solid liquid liquid water

## **Synthesis**

# $A + B \rightarrow AB$

- Two or more reactants make one product.
- Combine elements; Combine compounds; Elements and compounds combine.

$$2 Al + 3 Cl2 \rightarrow 2 AlCl3$$

$$P_4 + 10 F_2 \rightarrow 4 PF_5$$

$$2 FeCl2 + Cl2 \rightarrow 2 FeCl3$$

$$NH_3 + HCl \rightarrow NH_4Cl$$

## **Decomposition**

$$AB \rightarrow A + B$$

- One reactant breaks apart into two or more products.
- Elements are produced; Compounds are produced; Elements and compounds are produced.

$$2 \text{ ICl} \rightarrow I_2 + \text{Cl}_2$$

$$2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$$

$$\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$$

# **Demo: Elephant Toothpaste**

#### Reactant:

hydrogen peroxide

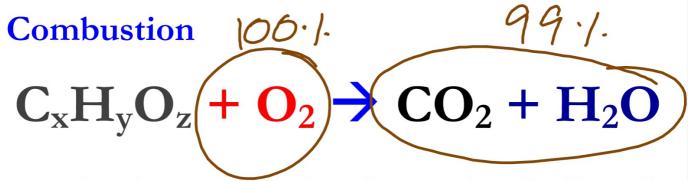
## **Products:**

water

oxygen gas

$$2H_{2}O_{2} \longrightarrow 2H_{2}O + O_{2}$$
 $H 4 \qquad H 4$ 
 $O 24 \qquad O 4$ 

 $2 H_2O_2 \rightarrow 2 H_2O + O_2$ 



- Requires oxygen and produces carbon dioxide and water.
- Typically produces heat and light.

$$CH_4 + 2 O_2 \rightarrow CO_2 + 2 H_2O$$
  
 $2 C_8H_{18} + 25 O_2 \rightarrow 16 CO_2 + 18 H_2O$   
 $2 CH_3OH + 3 O_2 \rightarrow 2 CO_2 + 4 H_2O$ 

## **Additional Notation**

$$(s) = solid$$

$$(1) = liquid$$

$$(g) = gas$$

$$(aq) = aquious$$

 $\Delta$  = heat or catalyst

 $\uparrow$  = gas or bubbles

 $\Psi$  = precipitate

 $\rightleftharpoons$  = reversible reaction

Pb (NO3)2 (aa)

H20(1)



# **Rules for Chemical Reactions:**

# Ago CO3

# **Predicting Products**

# **Predicting Products**

# Worksheet: Reaction Type and Balancing

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Exit Ticket #40:

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**Reaction Type**